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Methods for preparing perfluorinated [18F]-radiolabelled nitroimidazole derivatives for cellular hypoxia detection

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**METHODS FOR PREPARING PERFLUORINATED [ $^{18}\text{F}$ ]-RADIOLABELLED  
NITROIMIDAZOLE DERIVATIVES FOR CELLULAR HYPOXIA DETECTION**

The present invention relates to the field of chemical synthesis of radiolabelled  
5 bioactive compounds for the detection of specific targets present in the tissues or  
cells of a patient. More particularly, the present invention relates to the field of  
chemical synthesis of radiolabelled compounds to be used as indicators of tissue  
hypoxia. The present invention provides methods for preparing said type of  
compounds, as well as useful precursors in said synthesis and methods for  
10 preparing the same. The present invention also relates to methods of using these  
radiolabelled bioactive compounds for the detection of specific targets present in  
tissues of a patient. More particularly the present invention relates to the detection  
of tissue hypoxia in a patient using said radiolabelled compounds.

Cellular hypoxia is a typical feature in various physiopathological processes as  
15 frequent as malignant tumor development, heart disease, stroke, diabetes and  
wound healing. In malignant tumors, experimental and clinical evidences have  
shown that the hypoxic fraction may influence the malignant phenotype, the growth  
rate and may reduce the sensivity to ionizing radiation and chemotherapeutic  
agents. In head and neck lymph nodes, and in cervix carcinomas for example,  
20 tumour hypoxia is associated on an individual basis with a higher rate of local  
recurrence after radiotherapy. In stroke and heart infarct, it has been shown that  
the severity of the tissue function impairment critically depends on the location and  
the amount of ischemic tissue.

In this framework, accurate determination of the magnitude of tissue hypoxia has  
25 always been a focus of intensive research. Such information is of great value as a  
prognostic factor of the severity of the disease, and as a tool to select for  
alternative therapies and to monitor the response to therapeutic interventions. In  
oncology, recent development in microelectrode techniques has permitted to  
measure the oxygen partial pressure ( $\text{PO}_2$ ) in experimental as well as clinical  
30 tumors. This technique has greatly contributed to the current knowledge on the

influence of hypoxia in tumor physiopathology and response to treatment. Such a technique however has important limitations. The sensitivity of the method is far from being optimal in the range of  $PO_2$  values ( $<10$  mm Hg) of interest in oncology. It lacks specificity as it is also very much influenced by the amount of tissue necrosis around the microelectrode. Besides, it is an invasive and time consuming technique that will never be spread out in a routine clinical environment. In stroke and heart disease, there is no method to directly measure tissue hypoxia which can thus only be inferred from indirect measurements of tissue metabolism and vascularisation.

Tumor hypoxia is detected using hypoxia-binding chemical markers. These markers are nitroheterocyclic compounds which exhibit a particular metabolism under hypoxic cellular conditions, and hence can covalently bind to intracellular macromolecules (e.g. proteins, RNA, lipids and DNA). These reduced moieties trapped into hypoxic cells, can be detected by immunofluorescence on tissue section or by flow cytometry using for both techniques specific antibodies. Tagged with an appropriate radioactive isotope, these reduced moieties could also be detected by nuclear medicine techniques. Misonidazole is the prototype of hypoxia-binding chemical markers. More recently, a tri- and pentafluorinated nitroimidazole derivatives, designated EF3 and EF5, respectively, have been synthesized (US patent No. 5,540,908 in name of Koch). In comparison with misonidazole, these 2 compounds have several advantages. Both compounds have a more specific binding to hypoxic cells, and the binding does not depend on the intracellular level of reductase systems. In addition, fluorochrome-conjugated specific antibodies have been generated for both EF3 and EF5. Oxygen-dependent binding have been reported in various experimental systems such as EMT6 spheroids, EMT6 tumors and Morris 7777 rat tumors. EF5 has been very recently approved by American Authorities for human studies and a phase I trial is in progress in the USA. Although very sensitive and specific, determination of cellular hypoxia with EF5 or EF3 remains however invasive as it requires the use of tissue specimens.

The [ $^{18}\text{F}$ ] monofluorination of bioactive compounds is known; usually, the syntheses make use of the classical nucleophilic displacement of a leaving group with [ $^{18}\text{F}$ ] fluoride anion. In particular, the method has been applied to the preparation of [ $^{18}\text{F}$ ]-Fluoroethanidazole (1) and [ $^{18}\text{F}$ ]-Fluoromisonidazole (2), two members of the nitroimidazole family.

In organic synthesis, the direct and selective perfluorination ( $-\text{CF}_3$  and  $-\text{CF}_2-$ ) remains a difficult problem because the nucleophilic substitution strategy could not be efficiently applied (3). Solutions have been brought, using sulphurated precursors, in the case of poorly functionalized aromatic compounds (4, 5), and recently applied to their [ $^{18}\text{F}$ ] labelling (6). In the aliphatic series, only one precedent has been found, concerning the introduction of a  $\text{CF}_2$  group into non-functionalized alkyl chains (7). Similar introduction of a  $\text{CF}_3$  group has been recently developed in our laboratory and applied to the [ $^{18}\text{F}$ ] labelling of simple alkanes (8).

The present invention aims at developing and testing labelled bioactive compounds, more particularly [ $^{18}\text{F}$ ]-EF3 and [ $^{18}\text{F}$ ]-EF5 for *in vivo* detection of hypoxia. Such a method would permit measurements of both the hypoxic fraction and the distribution of hypoxia within an individual tissue or tumor. In comparison with the existing methods for measuring hypoxia (e.g., microelectrode, immunofluorescence and/or flow cytometry to detect hypoxia-binding chemical markers), the PET detection is a non-invasive technique that would allow individual measurements in any tumors and tissues. In comparison with other nuclear medicine techniques (e.g. SPECT), the PET camera detection offers the advantage of a better spatial resolution and a much more accurate quantification of the radioactivity. In comparison with other hypoxia-binding chemical markers, [ $^{18}\text{F}$ ]-EF3 and [ $^{18}\text{F}$ ]-EF5 would maintain both their superior specificity and sensivity for hypoxic cells as observed for the unlabelled parent compounds. Such a technique could be easily combined with anatomic imaging modalities (e.g. CT Scanner and MRI) allowing a better mapping of the distribution of hypoxia in a specific tissue/organ. In addition, the detection of hypoxia by the PET method could also be

combined with other functional imaging techniques (e.g. fMRI, PET with other markers) investigating important physiological parameters such as tissue proliferation or metabolism. Such a combined approach should allow to non-invasively study intriguing physiopathological questions related to tumor development and response to treatment, or to functional tissue defect after an ischemic injury. Important physiopathological questions related to tumor development and response to treatment, or to the understanding of functional tissue defect after an ischemic injury could be investigated by this nuclear medicine technique.

Thus assessment of tissue hypoxia with [ $^{18}\text{F}$ ]-EF3 or [ $^{18}\text{F}$ ]-EF5 is likely to allow significant benefits to the management of cancer and other human diseases. The structures of EF3 and EF5 are shown in Figure 1.

The present invention thus aims at methods for synthesizing perfluorinated radiolabelled bioactive compounds which selectively react with a target present in patient cells.

The present invention concerns the preparation of original chemical precursors allowing the direct radiolabelling of perfluoroalkyl groups ( $-\text{CF}_3$ ,  $-\text{CF}_2-$ ) by [ $^{18}\text{F}$ ] on substrates equipped with nitrogen-containing functions.

More particularly, the present invention aims at methods for synthesizing [ $^{18}\text{F}$ ] labelled perfluorinated nitroimidazole derivatives, more particularly methods for synthesizing [ $^{18}\text{F}$ ] labelled EF3 and EF5.

The present invention also aims at useful precursors for synthesizing said compounds and methods for preparing the same.

The present invention further relates to the different uses of said perfluorinated radiolabelled bioactive compounds, more particularly the different uses of [ $^{18}\text{F}$ ] labelled nitroimidazole derivatives, and even more particularly the uses of [ $^{18}\text{F}$ ] labelled EF3 and EF5.

According to a first aspect, the present invention relates to a first type of precursor compound which is a perfluorinated amino acid derivative which is N-protected by an imido group or a synthetically equivalent group and wherein the carboxyl



function has been transformed into a dithioester function or a synthetically equivalent persulphurated moiety.

Such persulphurated amino acid derivatives may be used to prepare perfluorinated amino acid derivatives using a suitable perfluorinating agent and a suitable oxidant as described below.

Said persulphurated compound may be derived from any of the following amino acids:  $\alpha$ -amino acids of the natural pool such as: glycine, alanine, valine, leucine, isoleucine, proline, phenylalanine, asparagine, glutamine, aspartate, glutamate, and bis-protected serine threonine, lysine, arginine, histidine;  $\alpha$ -amino acids of synthetic origin and derivatives thereof;  $\beta$ -amino acids such as 3-aminopropionic acid and derivatives thereof,  $\gamma$ -amino acids such as 4-aminobutyric acid and derivatives thereof,  $\delta$ -amino acids such as 5-aminovaleric acid and derivatives thereof,  $\epsilon$ -amino acids such as 6-aminocaproic and derivatives thereof and similarly, the  $\omega$ -amino acid derivatives.

Said imido group may be any imido group known in the art, such as for instance a phthalimido group.

Said first type of precursor compound according to claim 1 is used for the synthesis of a final precursor compound as described below. Figure 2 in combination with Figure 4 describes a representative synthesis of a precursor derived from  $\beta$ -alanine. Ethyl 3-(N-phthalimido)aminopropanedithioate having the general formula of the endproduct **(3)** of the reaction scheme presented in Figure 7 or Figure 2 is an example of a persulphurated beta-alanine derivative.

According to a preferred embodiment, the present invention relates to a method for preparing a compound of claim 1 as shown in Figure 7 and 8 comprising the following steps:

- (a) adding a THF solution of **2** to a suspension of PYBOP in THF followed by  $\text{Et}_3\text{N}$ ,
- (b) adding an amine **1** and  $\text{Et}_3\text{N}$  to the solution obtained in step (a),

- (c) adding a catalytic amount to the solution obtained in step (b) of pTsOH and refluxing the solution,
- (d) cooling the solution obtained after step (c) at ambient temperature and adding a sodium bicarbonate solution,
- 5 (e) extracting the product obtained after step (d) with ethyl acetate and drying and concentrating the product with ethyl acetate,
- (f) purifying the residue obtained after step (e) by column chromatography on silica gel,
- (g) removing traces of water by washing the product of step (f) with
- 10 trifluoroacetic anhydride, and,
- (h) reacting said persulphurated derivative obtained from step (g) with a suitable perfluorinating agent and a suitable oxidant resulting in a compound having a high yield of fluor atom incorporation.

Compound 3 as shown in Figure 7 is known as Ethyl-3-(N-phthalimido)aminopropanedithioate. Compound 1 and its synthesis have been

15 described in Josse et al., 1999. Preferably, said method is used to prepare namely N-(phthalimido)3,3,3-trifluoropropylamine having the general formula of the endproduct 4 of the reaction scheme presented in Figure 8 as described in Example 2.

20 Said perfluorinated precursor compound according to this embodiment of the present invention is used for the synthesis of a final precursor compound which is an amino synthon which can be incorporated in the synthesis of [ $^{18}\text{F}$ ] labelled target-bioactive compounds by using the classical methods of peptide coupling, or other coupling methods. Examples of suitable perfluorinating agents and suitable

25 oxidants according to the present invention are given in Figure 3. Details of the preparation process of the [ $^{18}\text{F}$ ] perfluorating agents appear in the earlier publications (6),(8).

According to a preferred embodiment, the present invention relates to a perfluorinated derivative compound of claim 2 obtainable by a reaction wherein

30 hydrogen fluoride/pyridine complex (HF-Pyridine) is used as a perfluorinating agent

and 1,3-dibromo-5,5-dimethylhydantoin (DBH) is used as an oxidant, resulting in a compound having a high yield of fluor atom incorporation. Said perfluorinated reagent and reaction product can contain [ $^{19}\text{F}$ ] or [ $^{18}\text{F}$ ].

According to another embodiment, the present invention also relates to mixtures of  
5 [ $^{19}\text{F}$ ] and [ $^{18}\text{F}$ ] labelled perfluorinated derivative compounds as defined above.

The [ $^{18}\text{F}$ ] isotope is incorporated in such an amount that the specific radioactivity of the compound is comprised between 1 and 30 Ci/mmol, preferably between 1 and 20 Ci/mmol, preferably between 1 and 10 Ci/mmol.

According to an even more preferred embodiment, the present invention relates to  
10 a compound having the formula of the endproduct 4 of the reaction as shown in Figure 8.

According to a preferred embodiment, the present invention relates to a perfluorinated derivative of the compound of claim 1 or 2 obtainable according to a reaction defined in any of claims 4 to 6, wherein radiolabelled [ $^{18}\text{F}$ ] is built in and  
15 the radiochemical yield of [ $^{18}\text{F}$ ] is incorporated in such an amount that the specific radioactivity of the compound is comprised between 1 and 30 Ci/mmol, preferably between 1 and 20 Ci/mmol, preferably 1 and 10 Ci-mmol.

According to another embodiment, the present invention relates to a final precursor compound which can be incorporated into the synthesis of [ $^{18}\text{F}$ ] target-bioactive  
20 compounds and which is a perfluorinated derivative of the first precursor described above wherein the nitrogen function has been deprotected by refluxing into hydrazine solution, resulting in a perfluoroalkyl amine derivative.

According to another embodiment, the present invention relates to a mixture of the radiolabelled perfluorinated bioactive compound and the non-radioactive labelled  
25 bioactive compound as defined above.

Preferably said final precursor according to the present invention is described in Figure 4 and is the endproduct (5). Since no procedure presently exists for direct and selective perfluorination of N-functionalized aliphatic compounds, the present invention brings a significant advance in organic chemistry in general (unlabelled

compounds), and in the [ $^{18}\text{F}$ ] radiolabelling of biologically active compounds in particular.

The method according to the present invention is flexible since [ $^{18}\text{F}$ ]-perfluorinated alkylamines can be used as building blocks in various total syntheses of pharmaceuticals. The method illustrated in the examples section can easily be expected to be extended to any other amino acid of interest.

According to a preferred embodiment, the present invention thus relates to an [ $^{18}\text{F}$ ] labelled bioactive compound synthesized using as a precursor a perfluorinated derivative according to claim 6 or 7.

According to a preferred embodiment, the present invention thus relates to the use of said perfluorinated derivative having the formula of the endproduct 5 of the reaction scheme as shown in Figure 4 for chemical synthesis of an [ $^{18}\text{F}$ ] labelled perfluorinated nitroimidazole having an incorporation of [ $^{18}\text{F}$ ] atoms in such an amount that the specific radioactivity of the compound is comprised between 1 and 10 Ci/mmol.

According to a preferred embodiment, said [ $^{18}\text{F}$ ] labelled perfluorinated nitroimidazole compound is [ $^{18}\text{F}$ ] labelled EF3 having a general formula as set out in Figure 1.

According to another preferred embodiment, said [ $^{18}\text{F}$ ] labelled perfluorinated nitroimidazole compound is [ $^{18}\text{F}$ ] labelled EF5 having a general formula as set out in Figure 1. [ $^{18}\text{F}$ ] labelled EF5 can be prepared by using an appropriate persulphurated precursor (see Figure 6 for potential precursor types).

The present invention also relates to a method for the detection of tissue hypoxia in a patient comprising:

- (a) introducing an [ $^{18}\text{F}$ ] labelled nitroimidazole compound as defined above into said patient,
- (b) imaging tissue hypoxia in said patient, and,
- (c) quantifying tissue hypoxia in said patient.

Said patient is preferably a mammal and more preferably a human. Preferred nitroimidazole compounds to be used according to this embodiment of the invention are [ $^{18}\text{F}$ ] labelled EF3 or [ $^{18}\text{F}$ ] labelled EF5.

5 Methods for detecting tissue hypoxia in patient tissue include, but are not limited to non-invasive imaging techniques, immunohistochemistry, immunofluorescence, autoradiography and flow cytometry. Imaging techniques include, but are not limited to positron emission tomography (PET). Generally, imaging techniques involve administering a compound with marker atoms which can be detected externally to the mammal. A compound of the invention is dissolved or dispersed in  
10 a pharmaceutically acceptable diluent, such as non-pyrogenic physiological saline, is administered to the patient preferably intravenously. After administration, time is allowed for metabolism (reduction) of the hypoxic marker and clearance of the non-metabolized compound. Tissue hypoxia is then assayed using one or several of the methods described above. Non-invasive imaging techniques can indeed be  
15 combined with immunohistochemistry, immunofluorescence, autoradiography or flow cytometry on tissue specimen.

According to a preferred embodiment, the detection technique used in said method is positron emission tomography.

The present invention also relates to a method for the detection of tissue hypoxia  
20 in a tissue comprising:

- (a) introducing an [ $^{18}\text{F}$ ] labelled nitroimidazole compound as defined above into a patient,
- (b) removing a tissue sample from said patient, and,
- (c) analysing the emission in said tissue sample by autoradiography.

25 Said patient is preferably a mammal and more preferably a human. Preferred nitroimidazole compounds to be used according to this embodiment of the invention are [ $^{18}\text{F}$ ] labelled EF3 or [ $^{18}\text{F}$ ] labelled EF5.

Also here, a compound of the invention, is dissolved or dispersed in a pharmaceutically acceptable diluent, such as non-pyrogenic physiological saline, is  
30 administered to the patient preferably intravenously. After administration time is

allowed for metabolism (reduction) of the hypoxic marker and clearance of the non-metabolized compound. A sample of for instance tumor tissue taken from the patient is then analyzed. Methods of obtaining tissue samples include any surgical and non-surgical technique known in the art. Surgical methods include, but are not limited to biopsy such as fine needle aspirate, core biopsy, dilation and curettage. According to another embodiment, the present invention relates to a method for the detection of [ $^{18}\text{F}$ ] labelled bioactive compound in a patient comprising:

(a) introducing an [ $^{18}\text{F}$ ] labelled bioactive compound according to claim 9 into said patient,

(b) imaging the presence of said [ $^{18}\text{F}$ ] labelled bioactive compound in said patient,

(c) quantifying the presence of said [ $^{18}\text{F}$ ] labelled bioactive compound in said patient.

Alternatively, the present invention also relates to a method for the detection of [ $^{18}\text{F}$ ] labelled bioactive compound in a tissue comprising:

(a) introducing an [ $^{18}\text{F}$ ] labelled perfluorinated nitroimidazole compound as defined above into a patient,

(b) taking a tissue sample from said patient, and,

(c) analysing the emission in said tissue sample by autoradiography.

Said patient is preferably a mammal and more preferably a human. Preferred nitroimidazole compounds to be used according to this embodiment of the invention are [ $^{18}\text{F}$ ] labelled EF3 or [ $^{18}\text{F}$ ] labelled EF5.

The examples as set out below are purely illustrative of a representative synthesis according to the embodiments of the present invention and are by no way intended to limit the present invention as set out in detail above. The content of all references referred to in this text is incorporated by reference.

**ABBREVIATIONS**

	EEDQ	N-Ethoxycarbonyl-2-ethoxy-1,2-dihydroquinoline
	PYBOP	Benzotriazole-1-yl-oxy-tris-pyrrolidino-phosphonium-hexa-
5		fluorophosphate
	THF	tetrahydrofuran
	pTsOH	para-toluenesulfonic acid
	Et <sub>3</sub> N	triethylamine
	DBH	dibromodimethylhydantoin
10	PET	Positron Emission Tomography
	MRI	Magnetic Resonance Imaging
	fMRI	functional Magnetic Resonance Imaging
	SPECT	Single Photon Emission Computed Tomography

### **FIGURE LEGENDS**

Figure 1 describes the chemical structure of EF5 and EF3

- 5 Figure 2 describes a reaction scheme for the synthesis of beta-Alanine Ethyl Dithioester. A part of this reaction scheme was previously described in Josse et al., 1999 (11).

- 10 Figure 3 describes the structure of the different perfluorinating agents and oxidants to be used in the labelling process according to the present invention.

Figure 4 describes the reaction scheme to prepare an [ $^{18}\text{F}$ ]-perfluoroalkyl amine derivative from the persulphurated first precursor obtained in Figure 2.

- 15 Figure 5 describes the synthesis of EF3. Radiolabelled EF3 is made by using [ $^{18}\text{F}$ ] perfluoroalkyl amine derivate of Figure 4 as a precursor in the last reaction step.

Figure 6 describes the different possible potential precursor types for the synthesis of EF5.

20

Figure 7 describes the reaction scheme for the synthesis of ethyl 3-(N-phthalimido)aminopropanedithioate as described in Example 1

- 25 Figure 8 describes the reaction scheme for the synthesis of N-(phthalimido)3,3,3-trifluoropropylamine as described in Example 2.



## EXAMPLES

### Example 1: Synthesis of ethyl 3-(N-phthalimido)aminopropanedithioate

5 A THF solution of **2** is added to a suspension of PYBOP in THF followed by  $\text{Et}_3\text{N}$ ; the mixture is stirred during 40 minutes at  $20^\circ\text{C}$ . Then, amine **1** (as the trifluoroacetate salt; (**11**)) and  $\text{Et}_3\text{N}$  are added, and the mixture is stirred during 3 h at  $20^\circ\text{C}$ . After this reaction time and the addition of a catalytic amount of pTsOH, the solution is refluxed overnight. After cooling at ambient temperature, a sodium  
10 bicarbonate solution is added and the product is extracted with ethyl acetate. Drying ( $\text{MgSO}_4$ ) and concentration under reduced pressure gave crude **3**. The residue is purified by column chromatography on silica gel (hexane/ethyl acetate 30:70). Last traces of water are removed by washing the product with trifluoroacetic anhydride. The total yield was 85%. A yellow solid product was  
15 obtained. Spectral data :  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 200 MHz)  $\delta$  1.26 (t, 3H,  $J=7.3$  Hz), 3.18 (q, 2H,  $J=7.3$  Hz), 3.37 (t, 2H,  $J=7$  Hz), 4.14 (t, 2H,  $J=7$  Hz), 7.73 (m, 2H), 7.85 (m, 2H);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 50 MHz) ppm 11.93, 30.70, 38.24, 49.13, 123.30, 131.99, 167.97, 233.00.

### Example 2: Synthesis of N-(phthalimido)3,3,3-trifluoropropylamine

The [ $^{18}\text{O}$ ]-water solution of [ $^{18}\text{F}$ ]-hydrogen fluoride is neutralized by a small amount of aqueous potassium hydroxide. Water is removed by evaporation till dryness, under an argon flux to give the potassium salt ([ $^{18}\text{F}$ ]-KF). Then, addition of a first  
25 portion of a dichloromethane solution of HF-Pyridine provides the desired radiolabelling agent. DBH is added and the mixture cooled to  $-78^\circ\text{C}$  before introducing **3**. The solution is allowed to reach the ambient temperature and is stirred for 30 minutes. A second fraction of a dichloromethane solution of HF-Pyridine is added for completing the reaction within 30 minutes. The trifluoromethyl  
30 amine **4** is recovered with a specific radioactivity comprised between 1 and 30

Ci/mmol, as measured by Radio-TLC (Radio-Thin Layer Chromatography).  $^{19}\text{F}$ -NMR (282 MHz)  $\delta$  - 66.2 (t,  $J$  = 10.5 Hz).

Example 3: Synthesis of 3,3,3-trifluoropropylamine

5

N-(Phthalimido) 3,3,3-trifluoropropylamine **4** is dissolved in acetonitrile and hydrazine hydrate (2 : 1), and heated at 75°C. The free amine is distilled under a slow stream of argon. The product is identified by comparison of the retention time in gas chromatography with authentic material.

10

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**CLAIMS**

1. A perfluorinated amino acid derivative which is N-protected by an imido group or a synthetically equivalent group and wherein the carboxyl function has been transformed into a dithioester function or a synthetically equivalent persulphurated moiety.
2. The perfluorinated amino acid derivative of claim 1 prepared by a method of synthesis comprising the steps of:
  - (a) adding a THF solution of 2 of Figure 7 to a suspension of PYBOP in THF followed by Et<sub>3</sub>N,
  - (b) adding an amine 1 of Figure 7 and Et<sub>3</sub>N to the solution obtained in step (a),
  - (c) adding a catalytic amount to the solution obtained in step (b) of pTsOH and refluxing the solution,
  - (d) cooling the solution obtained after step (c) at ambient temperature and adding a sodium bicarbonate solution,
  - (e) extracting the product obtained after step (d) with ethyl acetate and drying and concentrating the product with ethyl acetate,
  - (f) purifying the residue obtained after step (e) by column chromatography on silica gel,
  - (g) removing traces of water by washing the product of step (f) with trifluoroacetic anhydride, and,
  - (h) reacting said persulphurated derivative obtained from step (g) with a suitable perfluorinating agent and a suitable oxidant resulting in a compound having a high yield of fluor atom incorporation.
3. The perfluorinated derivative compound of claim 2 obtainable by a reaction wherein hydrogen fluoride/pyridine complex (HF-Pyridine) is used as a perfluorinating agent and 1,3-dibromo-5,5-dimethylhydantoin (DBH) is used as

an oxidant resulting in a compound having a high yield of fluor atom incorporation.

4. The perfluorinated compound according to claim 3 having the formula of the  
5 endproduct (4) of the reaction scheme as shown in Figure 8.

5. The perfluorinated derivative according to any of claims 1 to 4, wherein  
radiolabelled [ $^{18}\text{F}$ ] is built, and in such an amount that the specific radioactivity  
of the compound is comprised between 1 and 30 Ci/mmol, preferably between  
10 1 and 20 Ci/mmol, preferably between 1 and 10 Ci/mmol.

6. The perfluorinated derivative according to any of claims 1 to 5 wherein the  
nitrogen function has been deprotected, resulting in a perfluoroalkyl amine  
derivative.

7. The perfluorinated derivative according to claim 6 having the formula of the  
endproduct (5) of the reaction scheme as shown in Figure 4.

8. The use of a perfluorinated derivative according to claim 6 or 7 in the synthesis  
20 of [ $^{18}\text{F}$ ] labelled bioactive compounds using classical peptide coupling methods,  
or other coupling methods.

9. An [ $^{18}\text{F}$ ] labelled bioactive compound synthesized using as a precursor a  
perfluorinated derivative according to claim 6 or 7.

10. The compound of claim 9 which is an [ $^{18}\text{F}$ ] labelled perfluorinated nitroimidazole  
compound having an incorporation of [ $^{18}\text{F}$ ] atoms characterized by a specific  
radioactivity of the compound comprised between 1 and 30 Ci/mmol, preferably  
between 1 and 20 Ci/mmol, preferably 1 and 10 Ci/mmol.

11. The compound of claim 10 which is [ $^{18}\text{F}$ ] labelled EF3 having a general formula as set out in Figure 1.

5 12. The compound of claim 10 which is [ $^{18}\text{F}$ ] labelled EF5 having a general formula as set out in Figure 1.

13. A method for the detection of tissue hypoxia in a patient comprising:

- 10 (a) introducing an [ $^{18}\text{F}$ ] labelled nitroimidazole compound of any of claims 10 to 12 into said patient,  
(b) imaging tissue hypoxia in said patient, and,  
(c) quantifying tissue hypoxia in said patient.

15 14. A method according to claim 13 wherein the detection technique used in said method is positron emission tomography.

15. A method for the detection of tissue hypoxia in a tissue comprising:

- 20 (a) introducing an [ $^{18}\text{F}$ ] labelled nitroimidazole compound of any of claims 10 to 12 into a patient,  
(b) removing a tissue sample from said patient, and,  
(c) analysing the emission in said tissue sample by autoradiography.

16. A method for the detection of [ $^{18}\text{F}$ ] labelled bioactive compound in a patient comprising:

- 25 (a) introducing an [ $^{18}\text{F}$ ] labelled bioactive compound according to claim 9 into said patient,  
(b) imaging the presence of said [ $^{18}\text{F}$ ] labelled bioactive compound in said patient,  
(c) optionally, quantifying the presence of said [ $^{18}\text{F}$ ] labelled bioactive  
30 compound in said patient.

17. A method for the detection of [ $^{18}\text{F}$ ] labelled bioactive compound in a tissue comprising:

(a) introducing an [ $^{18}\text{F}$ ] labelled bioactive compound of claim 9 into a patient,  
(b) taking a tissue sample from said patient, and,

5 (c) analysing the emission in said tissue sample by autoradiography.

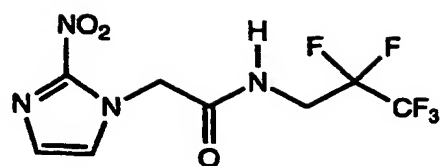
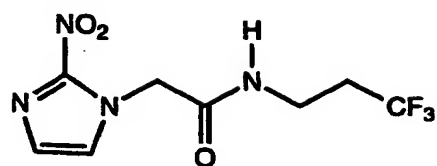


**ABSTRACT**

The present invention relates to chemical synthesis of radiolabelled perfluorinated bioactive compounds. More particularly, the present invention relates to radiolabelled compounds to be used as indicators for tissue hypoxia, More particularly, the present invention relates to the synthesis and the use of [ $^{18}\text{F}$ ] labelled perfluorinated nitroimidazole compounds having an incorporation of [ $^{18}\text{F}$ ] atoms characterized by a specific radioactivity of the compound comprised between 1 and 30 Ci/mmol, preferably between 1 and 20 Ci/mmol, preferably 1 and 10 Ci/mmol. More particularly to [ $^{18}\text{F}$ ] labelled EF3 or [ $^{18}\text{F}$ ] labelled EF5. The present invention also relates to a method for the detection of tissue hypoxia in a patient comprising introducing an [ $^{18}\text{F}$ ] labelled nitroimidazole compound into said patient, imaging tissue hypoxia in said patient, and, quantifying tissue hypoxia in said patient.

15



**EF5****EF3****Figure 1**

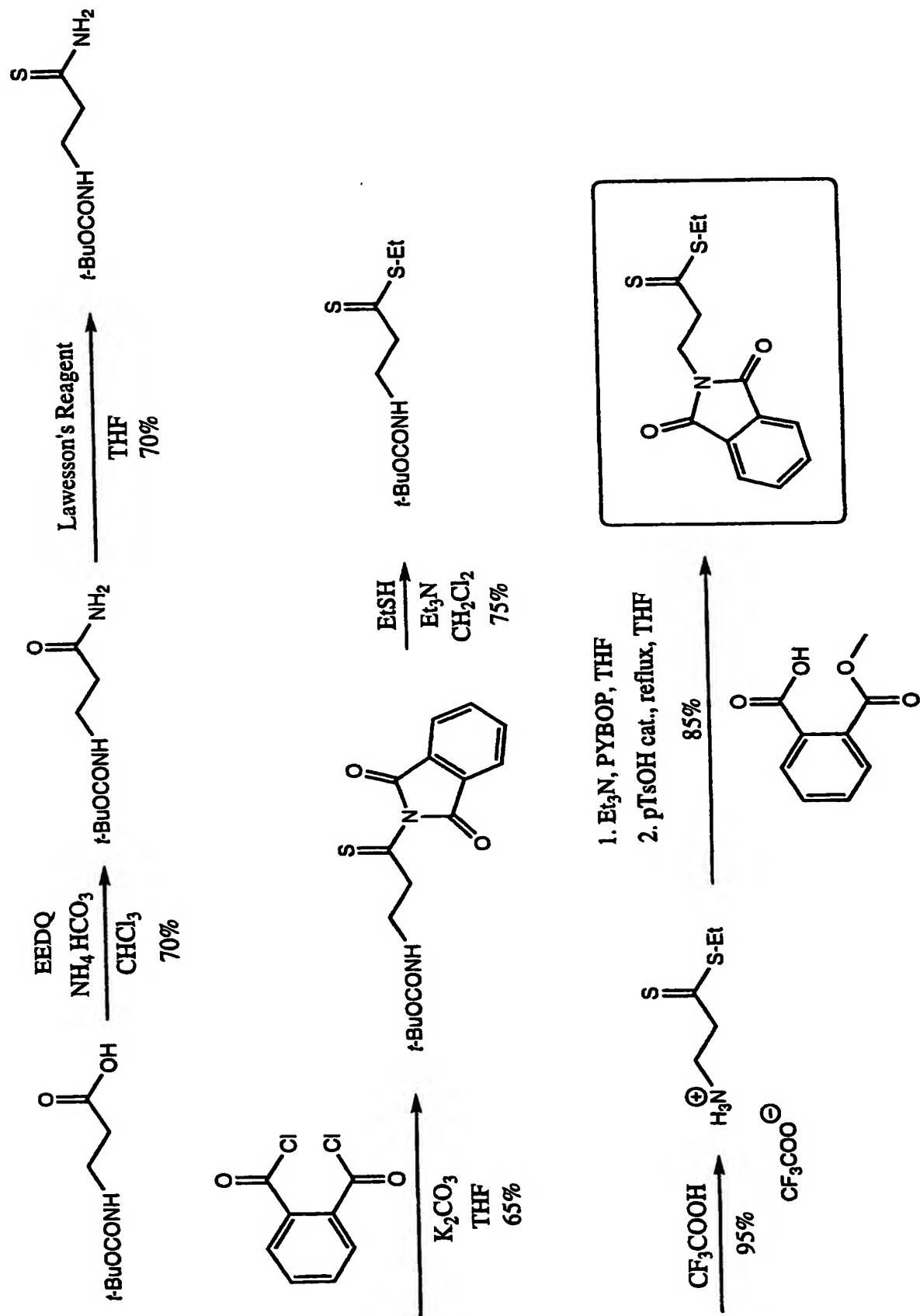
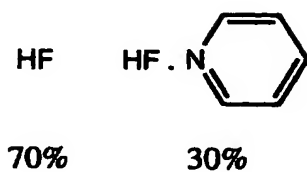
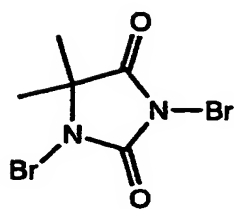
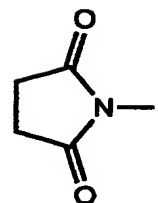
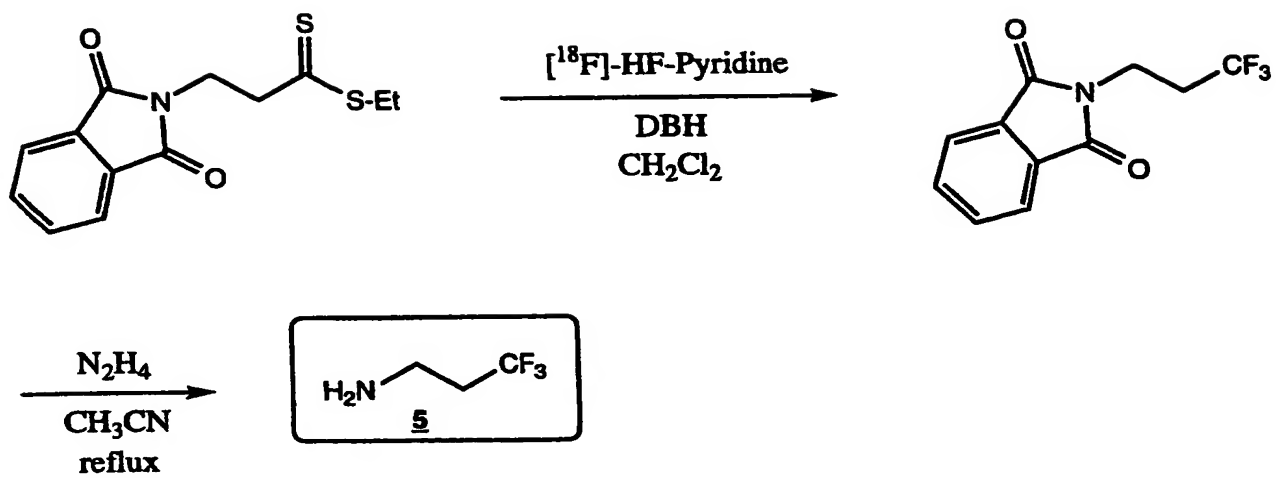
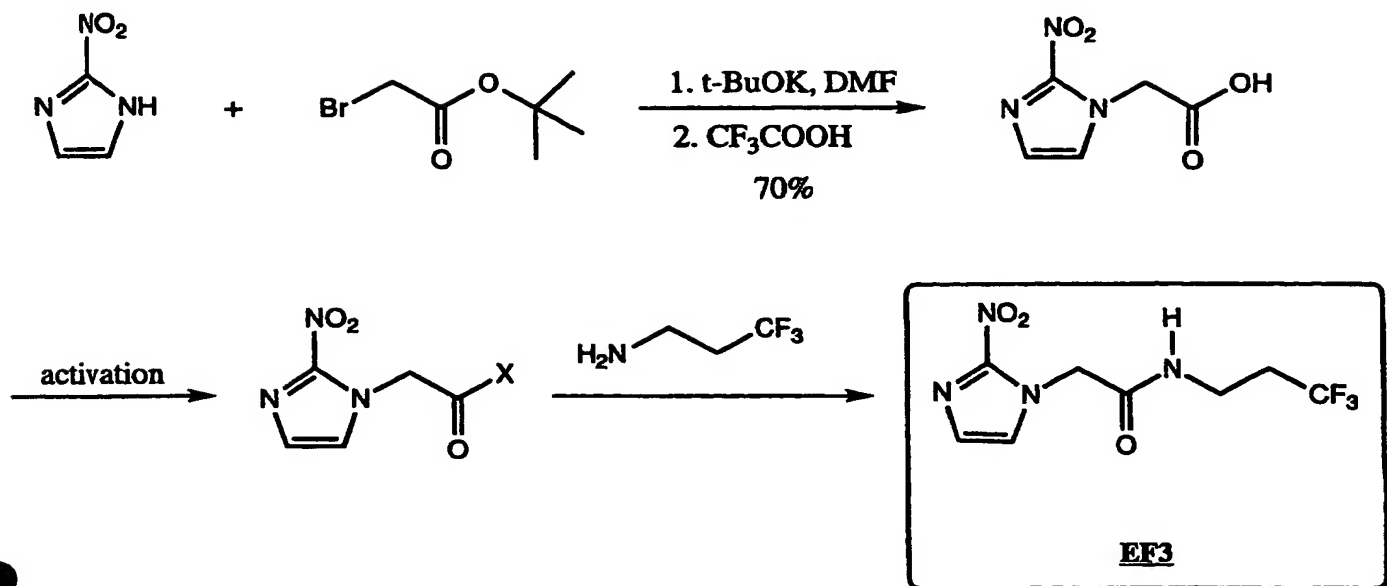
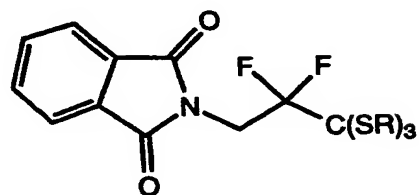
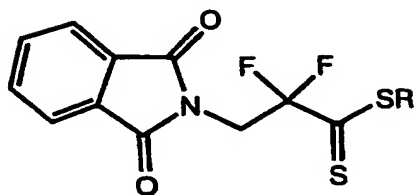
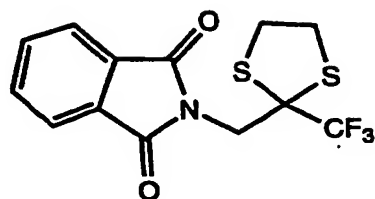
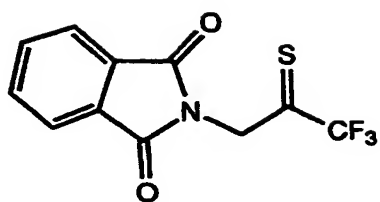


Figure 2

Perfluorinating agents**HF-Pyridine****TBAH<sub>2</sub>F<sub>3</sub>**Oxidants**DBH****NIS**Figure 3

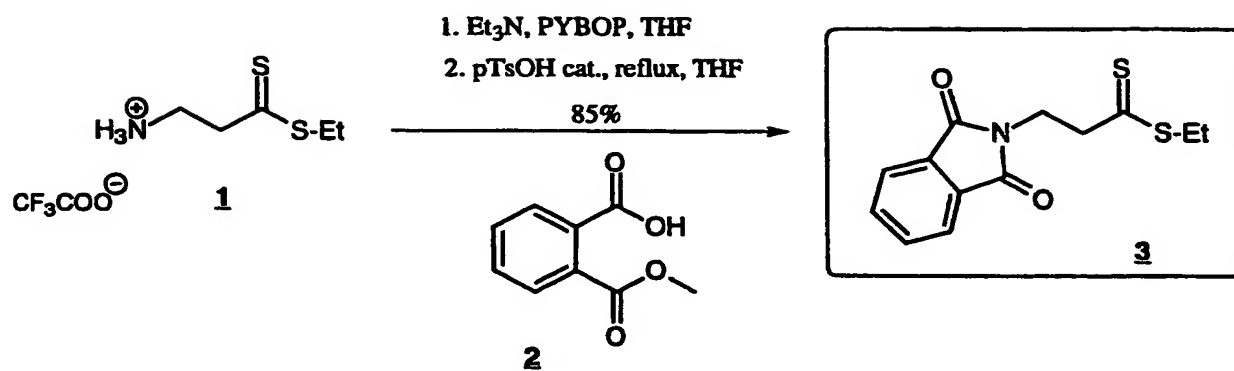
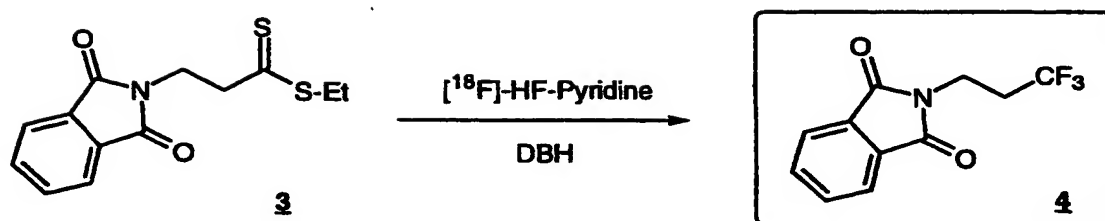
*Figure 4*

*Figure 5*



***Figure 6***



*Figure 7**Figure 8*

